

" Model Answer

INSTRUCTIONS:

- Check that the exam consists of (11) pages
- All questions are to be attempted
- o Answers should be written in the specified spaces
- o Blue pens should be used
- o Oral exam will be immediately after the end of the written exam
- Each student should commit to his/her oral exam committee
- Each student should assign in the attendance sheet
- Mobile phones shouldn't be hold

Good Luck

	DR. NAHLA ELASHMAWY
EXAMINERS	DR. HODA ELBAHRAWY
4	DR. EMAN GOUDA

18, 2014

BIOCHEMISTRY FINAL EXAM, SECOND YEAR STUDENTS, FIRST SEMESTER

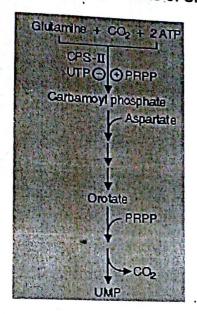
Answer Sheet for Question Two

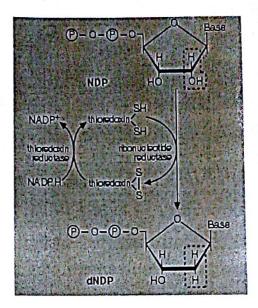
No. a	b	c	d	e	f		h	Not	a	b	Ċ.	d.	e '	f	g	h
1 2	1 Salar	P		4	The Park			20					441 4	\$2. Z.	2	b An
2 (29							201	
3								30			1. 3					
47.0		-						31	1			三				
5								32	افت-ق			10/0/20	4			
6			7.5					33; 34					- 11			_
.7								325		- T	_			€3.		
8								35 36	61.EN							
9			- =					37			1 THE 1		_			
10	通行							38	A 100 H		# 3					
11]								39	1							
12								40								
13/ L 14 . 15/ L								41		1					-	
14	75 5							. 42			- 14 14					
15		# =						43			= =				-	
16		ng tra						44			or annual contraction of the con	= :			-	
17		-						45			31. =1.			-	-	-
18	-							46			F-43		-	-	-	
19				11				47		Linear John			-		-	
20								48		-				-		-
21								49	The last live of the la					-	-	-
22	-				基	- 3		50	25				-			-
23	-					- Land		50 51	-	eter control	-	-	-	-		¥.
						-		-		10		-	-			
24	and the same							52				-	-	-		
25								53		-						
26							-	54				b.				
27								55								
18											Y WIN	ender were en			THE STATE OF THE S	

GOOD LUCK

Question One: Illustrate each of the following using equations or diagrams whenever possible: (5x4.5= 22.5 Marks, 25 min)

(1) The scheme of biosynthesis of UMP





- (2) The chemical reaction for conversion of ribonucleotides to deoxyribonucleotides
- (3) The reaction for biosynthesis of uric acid from hypoxanthine

Page 2 of 11

1

(4) Creatine kinase (CK) in a	
(4) Creatine kinase (CK) isoenzymes	15, FIRST SEMESTER
* Mya.Cardial Isoenzymes	
that contains 7 5% of total CK	1 1
III contains 75% of total of	ny tissue
The state of the s	>C.1.\2.(\4.15).
* Following an acute MI 2 CK as	pears
pain, reaches a peak of	f chest
pain, reaches a peak of	activity
at 24 hrs, and returns to be	an line
after 48 to 72 bec ou	1 2 2 2 2 2 1
after 48 to 72 hrs. exi	A. compared of the polypept of
CK MM () LAFO	X S talms:
(5) Effect of Linewaver-Burk Plot on noncompetitive inhibition	
max decreases in the presence	ond CK-BB
man-competitive inhibitor, whereas	K is
unchan a - 1" 1	-1
unchanged: 1 withinkib	, t.e. r
	no inhibitor
	7,10 7,7,74,5,7.4.
$\frac{1}{2}$	
max.	•
Vmax	1
Km	
Question Two: Select the ONE correct answer and	write the selected
letter in the provided answer sheet on page 1:	
	2.5 Marks, 60 min)
(1) In the beta-pleated sheets, the R-groups lie in between the a. True	e sheets.
(2) In sickle cell anemia, the amino acid glutamate in the β-gl	obin gene is replaced
by:	
a. Alanine (b) Valine . c. Aspartae d. leuci	ne

Are more dense than VLDL

c. Its major component is cholesteryl

esters

(13) Hydrolysis of chylomicrons by lipoprotein lipase produces:

- Chylomicron remnants
- Fatty acids

- Glycerol
- All of the above

(14) Cytosine can be spontaneously deaminated to give:

- a. Adenine
- Guanine

(b.) Uracil

(15) Nucleases act as:

- **Esterases**
- Phosphodiesterases

- Oxidases
- Reductases

Thymine

Select the amino acid (A-G) corresponding to each of the descriptions (16-25) and write the selected letter in the provided answer sheet on page 1:

H-CH-COOH

NH₂

H₂N-(CH₂)₄-CH-COOH NH₂

- (16) A semi-essential amino acid C
- Hydrophobic amino acid ____ A (17)
- (18) An amino acid without optical activity ___ B
- (19) Basic and aromatic amino acid ___ C
- (20) Constitutes about one third of collagen _ B
- Formed by post-translational modification ____ F (21)
- Forms covalent cross links in collagen ______ (22)
- Having no genetic code _____ F (23)
- (24)
- The smallest amino acid ___ B (25)

January 18, 2014

BIOCHEMISTRY FINAL EXAM, SECOND YEAR STUDENTS, FIRST SEMESTER

Select	the lipid substance (A-H) corres
C. E.	Cholesterol ester Choline phosphate D. Phopatidylinositol F. 2-Monoacyl glycerol
(26)	U. FIJOSDNATIDIC acid
(27)	Is produced from the hydrolysis of lecithin by phospholipase C — G
(28)	pancieatic lipase reaction
(29)	Is 1,2-diacylglycerol-3-P — H
(30)	Is diphosphatical and S
(31)	Is a phosphatic to the control of th
(32)	Is a phospholipid containing sugar D
(33)	Is the precursor for synthesis of prostaglandins ————————————————————————————————————
(34)	Can act in signal transduction
,	Is the precursor for synthesis of triacylglycerols in the intestinal mucosal cells
(35)	Its deficiency causes respiratory distress syndrome B
(37) Th	ne general formula of monosaccharides is: a) $C_n H_{2n}O_n$ b. $C_{2n} H_2O_n$ c. $C_n H_2 O_{2n}$ d. $C_n H_{2n}O_{2n}$ ne general formula of polysaccharides is: $(C_6 H_{10}O_5)_n$ b. $(C_6 H_{12}O_5)_n$ c. $(C_6 H_{12}O_6)_n$ d. $(C_6 H_{10}O_6)_n$
(38) Co co a c.	ompounds having the same structural formula but differing in spatial stereoisomers Optical isomers Observed by the same structural formula but differing in spatial but d
(39) The (a.)2	number of epimers of glucose is:
(40) Hya (a) J	oints b. Brain c. Abdomen d. Mouth
a. G	rose consists of: lucose + Glucose lucose + Galactose b. Glucose + Fructose

d. Glucose + Mannose

c. Glucose + Galactose

(42) A carbohydrate found only in milk is:

- a. Glucose
- b. Galactose
- (C)Lactose
- d. Maltose

(43) In the Michaelis-Menten equation:

- a. K_m is the amount of substrate which gives maximum catalytic activity
- b. V_m is the half maximum rate of reaction
- c. V_{m} is the amount of enzyme required to achieve maximum velocity
- d.) K_m is the substrate concentration that will give 1/2V_{max}

(44) Dependence of reaction velocity on substrate concentration is described as:

- a. Equal to K_m
- b. Independent of enzyme concentration
- (c) Proportional to the amount of ES complex
- d. Zero order with respect to substrate

(45) Which of the following describes a characteristic of most allosteric enzymes?

- a. They are composed of single subunits.
- b. They generally follow Michaelis-Menten kinetics.
- (c.) They show cooperativity in substrate binding.
- d. They have irreversible allosteric inhibitors that bind at allosteric sites.

(46) Urease catalyzes the hydrolysis of urea but not of diethylurea, because:

- a. It has relative specificity
- (b.) It has absolute specificity
- c. It has broad specificity
- d. None of the above

(47) Which one of the following statements is FALSE for the allosteric enzyme?

- a. The binding of a positive allosteric effector
- (b.) They frequently catalyze a committed step in increases K_m a metabolic pathway
- c. They are often composed of more than one subunit
- d. They do not follow Michaelis-Menten kinetics

(48) The presence of a non competitive inhibitor leads to:

- a. An increase in the V_{max} & K_m
- (b) A decrease in the V_{max} without affecting K_m
- c. A decrease in K_m & V_{max}
- d. An increase in K_m without affecting V_{max}

(49) In the presence of the allosteric activator, the V_{max} for the enzyme would

(a) Increase

b. Decrease

BIOCHEMISTRY FINAL EXAM, SECOND YEAR STUDENTS, FIRST SEMESTER

(50) Which of the following Linewearver-Burk plot?	pairs cannot be	distinguished	on	the b	asis of
- Dark blots					

- a. Competitive-Noncompetitive
- (c) Noncompetitive-Irreversible
- b. Competitive-Irreversible
- d.Noncompetitive-Uncompetitive

(51) Metal cations may do all of the following in metallo-enzymes EXCEPT:

- a. Participate in oxidation-reduction processes
- (b.)Act as uncompetitive inhibitors
- c. Stabilize the active conformation of an enzyme
- d. Form chelates with the substrate

(52) A fixed amount of a chemical agent reduces the catalytic activity of lactate dehydrogenase, with a decrease in V_{max} . The K_{m} is unaffected. This inhibitor is:

- a. Definitely a competitive inhibitor
- b.)Definitely a non competitive inhibitor
- c. Definitely an irreversible inhibitor
- d. Either a competitive or an irreversible inhibitor

(53) Types of physiological regulation of enzyme activity do not include:

a. Covalent modification

or idea the for

- (b.) Irreversible inhibition
- c. Competitive inhibition

d. Allosteric activation

(54) How many different proteins are present in normal LDH?

THE THE STEAM OF

- a. One
- (b.)Two
- c. Three
- d. Four
- e. Five

CH₃-CH-CH₂-COO CH3-C-CH2-COOT

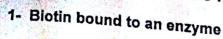
(55) Conversion of β-hydroxybutyrate to acetoacetate (shown above) occurs by:

- a.) Oxidation
- b. Reduction
- c. Dehydration
- d. Dehydroxylation

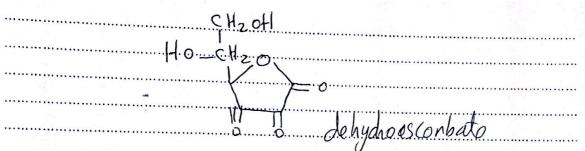
correct the false ones:
1-Protoporphyrin IX contains (10 marks, 10 min)
1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
2-VitaminD opposes the effect of parathyroid hormone in regulation of blood calcium
. P. arathy. roid hormone inchase the love of vitamin of to 1 Catal
3-Anticonvulsant drugs increase ALA synthase activity
4-Zinc is a trace at-
4-Zinc is a trace element which is essential for the synthesis of vitamin A transport protein (RBP)
5-Thiamine deficiency results in a decreased production of ATP and thus impaired cellular function ()
6-The affinity of hemoglobin for the first oxygen bound is approximately 300 times greater than its affinity for the last oxygen bound (F) The affinity of Hb for the last oxygen bound is approximately 300 times (F) 7 Sulfanamida is compatitively inhibiting the cyathesia of folia acid in 15th Ca bound
7-Sunonarride is competitively inhibiting the synthesis of folic acid in microorganisms
and human and thereby decreases the synthesis of nucleotide for replication ()
and human and thereby decreases the synthesis of nucleotide for replication (+) Folic. aud. is. not. Synthesis. in human. only in microorg
8-Calcitriol is a potent ligand of the vitamin D receptor which mediates most of actions of the vitamin
8-Calcitriol is a potent ligand of the vitamin D receptor which mediates most of actions of the vitamin 9-Retinol is the light –absorbing portion of rhodopsin and can be obtained from retinoic
8-Calcitriol is a potent ligand of the vitamin D receptor which mediates most of actions of the vitamin 9-Retinol is the light –absorbing portion of rhodopsin and can be obtained from retinoic
8-Calcitriol is a potent ligand of the vitamin D receptor which mediates most of actions of the vitamin 9-Retinol is the light –absorbing portion of rhodopsin and can be obtained from retinoic

Question Four: Draw the following structure

(10 marks, 10 min)



2- Dehydroascorbate



3- Pyridoxal phosphate

Micotinamide adenine dinucleotide

$$O = P = 0$$
 $O = P = 0$
 $O =$

Question Five: Write briefly on each of the following using diagrams or figures whenever possible: (5x5= 25 marks, 15 min)

1- Blosynthesis of heme and its regulation.
111111111111111111111111111111111111111
Enzymp, ALA synthase J step () Hemp, Color PLP S-Anima leveling and (A) A)
En Zua Ala III S-Ammo levulinicació (ALA) A GO
3 Polis ALA dehydratase step 2/ /ead femochdotal Iron
2- Role of Vitamin K as clotting vitamin. Porphobilingen further protoporphyrin
MACH-SIMMATURE CONTRACTOR
Colnz. PLP S-Animo levulinic acid (ALA) from Enzyme: ALA dehydratase step 26 lead ferrochdatus Iron 2-Role of vitamin K as clotting vitamin. Porphobilinogen further protoporphyrin NAM CH- CH- Charles I VIII IX, X warfurin (a) -> 0 Manfurin (a) -> 0 Manf
3- Role of TPP as-a coenzyme in the formation or degradation of ketols.
3- Role of IPP as-a coenzyme in the formation or degradation of ketols.
Trans Cataloga I N Value of Ca
Thiampp Solution
Transketolese > Xylulose 5-p Thianump P Sedoheptulose 5-p Glucos delabores > 0
4- Vitamin D deficiency
a Nutritional Rickets: Vitamin Doleficiency causes rickets in
children 30 steomalacia in adults characterized by soft plieble
bones 31 bones susceptibility to a cotinging of the pheble
E) lend hickets: chronic vana fail und abilit + 1
Dones 3.1 hones susceptibility to fracture in adults. E) lend rickets: chronic vene I failure & ability to form the 5-Daily energy requirements. active form of vit.D.
estimated by measurment of energy Expenditure
Daily energy Expenditure Basal metabolic vate + specific
dynamic action (SDA) + physical activity
BMR = weight (kg) x 24 Kcal/day
physical activity sadatas your
physical activity; > Sedentary 30% BMR GOOD ENCK
/ Si Moderate 40/BMR
Page 13 - Can
Heavy Soj. BMR